

Phys 448 HW 10

- 1) BD 9.1
- 2) Suppose that on one of the sites (#2) of a benzene molecule (BD 7.10) an "impurity" ^{14}C atom replaces the normal ^{12}C atom. The atoms are nearly chemically identical, but the change in mass causes the distance between the ^{14}C and its neighbors to be slightly different. This in turn implies that the tunneling of an electron between the ^{14}C and its neighbors is slightly different, *i.e.* the coupling changes as $A \rightarrow A + b$. We wish to calculate the effect of the change in the tunneling on the energy levels using perturbation theory. First, divide the Hamiltonian into the H_0 of BD 7.10 and a perturbation H_1 . Write these matrices in the $\{|\xi_n\rangle\}$ basis. As usual, find the eigenvectors and eigenvalues of the unperturbed Hamiltonian using the CSCO of H_0 and R . In Mathematica, defining a function
$$\text{toPolar}[x_]:=x/.a_ \rightarrow \text{Abs}[a] \text{Exp}[I \text{Arg}[a]]$$
will convert ugly complex numbers into a nicer polar form.
- 3) Use first order perturbation theory to find the energy shift of the lowest and highest energy level of the benzene molecule due to the impurity.
- 4) The first two pairs of excited states of the molecule are two-fold degenerate, so degenerate perturbation theory is required to find their energy shifts. Find the effective 2-level Hamiltonians for each degenerate pair, using the CSCO basis, and find the resulting energies.
- 5) Plot the energies from perturbation theory of each of the states as a function of $-1 < \frac{b}{A} < 1$. Also find the exact energies and plot them on the same plot. Comment on the range over which perturbation theory gives good results.
- 6) A three-level quantum system has energies $\{A, 0, -A/2\}$. A perturbation is added: $V = b(|1\rangle\langle 2| + |2\rangle\langle 3|) + \text{h.c.}$ where h.c. denotes Hermitian conjugate. Find the energy shifts to second order.